

## Electron Dot Practice

Draw the electron dot structures for each of the following:

$\text{CCl}_4$	$\text{HCl}$
$\text{C}_2\text{H}_6$	$\text{H}_2\text{O}$
$\text{BH}_3$	$\text{NH}_3$
$\text{HF}$	$\text{C}_2\text{H}_2$
$\text{PCl}_3$	$\text{N}_2\text{H}_2$
$\text{PBr}_3$	$\text{CO}_2$
$\text{N}_2$	$\text{CCl}_2\text{F}_2$